



Chemistry CURRICULUM OVERVIEW 2023/2024

Week	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38
Year/Term	Autumn											Spring											Summer															
SEVEN	Starting Science Safety Investigation Graphs	Cells Plant and animal cells Using microscopes	Changing States Solid, Liquids and Gases Particles States of matter BP/MPs	Energy transfers Types of energy Transfers and transformations	Nutrition Balanced Diet, Food groups, Food tests	Separating mixtures Purification of solutions	Energy Resources Energy as a resource, Diverse energy resources	Adaptations Adaptations Evolution	Forces Force diagrams Types of force	Rocks and weathering Types of rock Rock cycle Earth structure Fossils	Ecology Variation, Diversity of species, Interdependence	Space Solar system Planetary motion																										
EIGHT	Elements, compounds, fuels Conservation of mass Formula equations Fuels Pollution	Light and sound White and coloured light Reflection/ray diagrams Pitch/frequency Human ear	Health and variation Microbes Disease Variation Natural selection	Respiration, photosynthesis, movement, Circulation, Breathing, Respiration, Photosynthesis, Musculoskeletal system, Exercise & Smoking	Electricity and magnetism Circuit diagrams Linear and parallel Electrical safety Magnets Electromagnets	Acids, alkalis, metals pH scale Neutralisation Reactions of metals Rust Making salts																																

Bold – topic headings
Metals acids, bases and salts
Atoms, ions, redox etc
Atmospheric chemistry
Chemical Analysis

NINE	Separating substances and water. Elements, mixtures, compounds Chromatography Water treatment Required practicals	Atomic structure and periodic table intro Atomic model Arrangement of the elements in the periodic table	Metals and reactivity Metals properties Extraction based on reactivity Redox in terms of oxygen	The atmosphere Evolution of the atmosphere Pollution sources	Rates of reaction Collision theory Factors effecting rate – concentration, surface area, temperature, catalysts Required practicals	
TEN	Structure and bonding History of atomic structure Isotopes States of matter Ions and ionic bonding Covalent bonding Metallic bonding	Oil (organic chemistry) Fractional distillation Simple molecules Alkanes, alkenes, polymers Chemical test Combustion and atmospheric chemistry	Periodic table group 1 metal reactivity Displacement and redox reactions of group 7 in terms of electrons Group 0	Electrolysis Ions and ionic compounds Redox Half equations Required practical	Acids ,bases and salts Acids as proton donors, reactions with metals, bases and carbonates Salt formation required practical	Quantitative chemistry Calculations involving mass, Mr, moles and concentration Titration required practical
ELEVEN	Exothermic, endothermic Required practical equilibria LCA	Rates Chemical analysis - (and for combined groups revision 5.1-5.4)	Triple – further organic Combined – revision 5.5-5.10	Revision		
TWELVE	Atomic structure including mass spectrometry Periodicity Group 2, Redox	Group 7, Amount of substance Kinetics Introduction to organic – nomenclature, isomers	Amount of substance Alkanes, alkenes, alcohol	Energetics equilibria Alcohol, Halogenoalkanes	Equilibria KSAS Organic synthesis IR Revision	Thermodynamics Carboxylic acids aspirin prep

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	Introduction to Chem Structure and bonding					
THIRTEEN	Entropy Acids and bases Optical isomers, aldehydes and ketones Benzene	Acids and bases Redox and electrochemistry Benzene Kinetics Kp Amides and amines	Transition metals Polymers Amino acids chromatography	Mocks transition metals periodicity DNA NMR organic synthesis	Required practical catch up and revision Organic synthesis and revision	

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